**A. N. O. Dodoo,<sup>1</sup> S. Bansal,<sup>1</sup> D. J. Barlow,<sup>1,2</sup> on alveolar absorption.<br><b>F. C. Bennet.<sup>1</sup> R. C. Hider.<sup>1</sup> A. B. Lansley.<sup>1</sup> To facilitate their preparation, detection and quantitation, F. C. Bennet,<sup>1</sup> R. C. Hider,<sup>1</sup> A. B. Lansley,<sup>1</sup>** 

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*Purpose.* To determine how the structures of peptides influence their alveolar permeability.

*Methods.* The studies were performed using 14 synthetic 'model' peptides, labelled with a novel, non-intrusive amino acid fluorophore, and their transport studied using rat alveolar cell monolayers cultured on permeable supports.

*Results.* The passage of the peptides across the epithelial cell monolayers is shown to be primarily paracellular, with an inverse dependence on molecular size, and an enhanced flux observed for cationic peptides. This label has previously been shown to be both chemically<br>The apparent permeability coefficients ( $P_{app}$ ) for the peptides (together and metabolical The apparent permeability coefficients ( $P_{app}$ ) for the peptides (together and metabolically stable, allowing the detection of peptides at with those for other organic solutes, taken from the literature) are shown to be w

*Conclusions.* The various monolayer  $P_{app}$  values correlate well with the results from *in vivo* transport experiments, and the conclusion is drawn that the pulmonary delivery of peptide drugs is perfectly **MATERIALS AND METHODS** exploitable.

**KEY WORDS:** transport of peptides; molecular structure; molecular **Peptide Syntheses** size; cultured alveolar cell monolayers; alveolar permeability.

as to the potential to be afforded by using the pulmonary route out using 2-(1H-benzotriazole-1-yl)-1,1,3,3-tetramethyluro-<br>of administration for the systemic delivery of protein and pep-<br>nium tetrafluoroborate in the pres tide drugs (1,2). There are no *systematic, comprehensive* studies ine. Peptides were cleaved from the resin using TFA, water yet reported, however, that demonstrate that the predicted poten- and triisopropylsilane (95:4.5:0.5). tial is both real and achievable, although there have been several studies concerned with the alveolar permeability of specific **Synthesis of the Peptide Label** peptides  $(cf., 3-5)$ . In the following work we have endeavoured

stable, and had no propensity to form regular secondary struc-<br>tures. It was also deemed important to use peptides possessing arin, and subsequent acid hydrolysis gave the novel amino<br>a reasonable aqueous solubility and, a a lack of pharmacological activity.

**Systematic Investigations of the** The three series of model peptides described in this study are based on the sequence triplet Gly-Ala-Ser and are presented **Influence of Molecular Structure on** in Table 1. Series I peptides were designed for the purpose of **the Transport of Peptides Across** exploring the relation between alveolar absorption and peptide<br> **Cultured Alveolar Cell Monolayers** the effects on alveolar absorption of modest (single residue) changes in hydrophobicity. Series III peptides were designed to permit an investigation of the influence of peptide charge

**M. J. Lawrence,<sup>1</sup> and C. Marriott<sup>1</sup> and C. Marriott<sup>1</sup> all of the model peptides were synthesized incorporating the** novel fluorescent amino acid *DL*-(6,7-dimethoxy-4-coumaryl) alanine (DCA, single letter code  $X^*$ ):



the monolayer tri-junctional complexes, and the latter are taken as<br>monolayer defects.<br>tigated using primary cultures of rat alveolar cells (nominally,<br>*Conclusions*. The various monolayer  $P_{\text{max}}$  values correlate well w

Peptide synthesis was carried out using Fmoc chemistry. on a Perseptive Biosystems 9050 Pepsynthesizer. A PEG-poly- **INTRODUCTION** styrene resin was used with a modified Rink amide linker. Fmoc Over the last five years or so there have been many musings amino acids were used in 4-fold excess and activation carried as to the potential to be afforded by using the pulmonary route out using 2-(1H-benzotriazole-1-vl)nium tetrafluoroborate in the presence of diisopropyl ethylam-

peptides (cf., 3–5). In the following work we have endeavoured<br>to rectify this deficiency, undertaking a systematic investigation<br>of the alveolar cell permeability of 'model' peptides with<br>defined physico-chemical characte ensure that the compounds were chemically and metabolically arin. Diethylacetamidomalonate was treated with sodium<br>stable and had no propensity to form regular secondary structure hydride followed by the 4-chloromethyl-6,

# **Transport Studies**

Alveolar cell monolayers were prepared according to Kim <sup>1</sup> Department of Pharmacy, Franklin Wilkins Building, King's College *et al.* (8). Alveolar cells were isolated from specific pathogen-London, 150 Stamford Street, London SE1 8WA, England. free Sprague-Dawley male rats by digestion with porcine pan-<br><sup>2</sup> To whom correspondence should be addressed. can be reatic elastase (2 U/ml; Worthington Biochemicals, N creatic elastase (2 U/ml; Worthington Biochemicals, NJ) and

**Table 1.** Physicochemical and Transport Data for Model Peptides

	Peptide	MW	$r/\AA$	logP	$N_h$	$P_{app} \times 10^{7} / \text{cm} \cdot \text{s}^{-1}$
Series I	$acX^*NH_2$	334	4.5	$-0.13 \pm 0.01$	12	$1.88 \pm 0.07$ (n = 3) (AB)
	$acX^*ASNH_2$	491	5.2	$-1.25 \pm 0.01$	20	$2.06 \pm 0.09$ (n = 3) (BA) $1.00 \pm 0.27$ (n = 17) (AB)
	acX*ASGASNH <sub>2</sub>	707	5.9	$-2.51 \pm 0.02$	31	$1.04 \pm 0.14$ (n = 8) (BA) $0.89 \pm 0.19$ (n = 4) (AB)
						$0.65 \pm 0.37$ (n = 6) (BA)
	$acX^*AS(GAS)_{2}NH_2$	922	6.4	$-3.43 \pm 0.06$	42	$0.50 \pm 0.24$ (n = 6) (AB) $0.69 \pm 0.13$ (n = 4) (BA)
	$acX^*AS(GAS)_{3}NH_2$	1137	6.9	$-3.22 \pm 0.03$	53	$0.60 \pm 0.11$ (n = 5) (AB)
	$acX^*AS(GAS)7NH2$	1998	8.3	$-2.99 \pm 0.05$	97	$0.41 \pm 0.26$ (n = 8) (AB) $0.38 \pm 0.11$ (n = 3) (BA)
Series II	$acX^*ASNH_2$	491	5.2	$-1.25 \pm 0.01$	20	$1.00 \pm 0.27$ (n = 17) (AB) $1.04 \pm 0.14$ (n = 8) (BA)
	$acX^*AVNH_2$	503	5.3	$-0.13 \pm 0.02$	18	$1.26 \pm 0.27$ (n = 4) (AB)
	acX*ASGASNH <sub>2</sub>	707	5.9	$-2.51 \pm 0.02$	31	$0.89 \pm 0.19$ (n = 4) (AB) $0.65 \pm 0.37$ (n = 6) (BA)
	$acX*ASGAVNH2$	719	6.0	$-1.54 \pm 0.03$	29	$0.98 \pm 0.15$ (n = 3) (AB)
	$acX^*AS(GAS)_{2}NH_2$	922	6.4	$-3.43 \pm 0.06$	42	$0.50 \pm 0.24$ (n = 6) (AB) $0.69 \pm 0.13$ (n = 4) (BA)
	$acX^*AV(GAS)_{2}NH_{2}$	934	6.5	$-2.76 \pm 0.05$	40	$0.66 \pm 0.03$ (n = 3) (AB)
Series III	$acX^*AS(GAS)_{2}NH_2$	922	6.4	$-3.43 \pm 0.06$	42	$0.59 \pm 0.08$ (n = 3) (BA) $0.50 \pm 0.24$ (n = 6) (AB) $0.69 \pm 0.13$ (n = 4) (BA)
	$X^*AS(GAS)_{2}NH_2$	880	6.4	$\le -3.5$	42	$0.84 \pm 0.04$ (n = 4) (AB)
	$acX^*AS(GAS)$	923	6.4	$<-3.5$	42	$0.82 \pm 0.16$ (n = 8) (AB)
	$acX^*AS(GAD)_{2}NH_2$	978	6.5	$<-3.5$	46	$0.48 \pm 0.16$ (n = 4) (AB)
	acX*ASGADGAKNH2	991	6.6	$<-3.5$	45	$0.65 \pm 0.35$ (n = 6) (AB)
	$acX^*AS(GAK)_{2}NH_2$	1004	6.7	$<-3.5$	44	$1.40 \pm 0.31$ (n = 3) (AB)

*Note:* Molecular weight (MW); molecular radius (r); log(octanol:water partition coefficient)(tabulated as logP, but for the series III peptides actually representing  $logD_{7,4}$ ); no. of potential hydrogen bonds formed with solvent water  $(N_h)$ ; apparent permeability coefficient for transport of the peptides across rat alveolar cell monolayers, in the apical-to-basolateral (AB) and basolateral-to-apical (BA) directions (values representing the mean  $\pm$  standard deviation, calculated over n experiments). Amino acids shown using standard single letter code. X\*, DL-(6,7-dimethoxy-4-coumaryl) alanine; ac, N-terminal acetyl group; NH<sub>2</sub>, C-terminal amide group.

were cultured in a humidified atmosphere of 5% CO<sub>2</sub>/95% air giving a final concentration of peptide in the donor medium of at 37°C, on tissue-culture treated polycarbonate filters (1.13  $\sim$  0.5 mM. At selected time inte at  $37^{\circ}$ C, on tissue-culture treated polycarbonate filters (1.13  $\text{cm}^2$ ; Transwell<sup>®</sup>, Costar, MA), and seeded (day 0) at a density of  $1.5 \times 10^6$  cells per cm<sup>2</sup>. The cells were cultured in minimum essential medium supplemented with 10% v/v newborn bovine replaced with equal volumes of fresh transport medium. At serum, 2 mM *L*-glutamate, 100 U/mL penicillin, 100 ng/mL the end of each transport experiment the donor and receptor streptomycin and  $0.1 \mu$ M dexamethasone. The medium on both compartment media were routinely assayed following separa-

Ringer's solution (transport medium) comprising 1.8 mM day 2 using a Millicell® ERS system (Millicell, MA). CaCl<sub>2</sub>, 0.81 mM MgSO<sub>4</sub>, 5.4 mM KCl, 116.4 mM NaCl, 0.782 For all of the transport experiments performed, the integrity serum albumin (fraction V), 25 mM NaHCO<sub>3</sub> in 15 mM HEPES the lial electrical resistance (*TER*) and apparent permeability buffer pH 7.4. The cell monolayers were washed twice with coefficient ( $P_{app}$ ) for <sup>14</sup>C-mannitol. buffer pH 7.4. The cell monolayers were washed twice with transport medium and the apical and basolateral compartments Apparent permeability coefficients for the various solutes filled with 0.6 mL and 1.5 mL transport medium, respectively. The cells were equilibrated with transport medium for 2 h at  $37^{\circ}$ C in a humidified 5% CO<sub>2</sub> incubator. Approximately 1 mM stock solutions of the model peptides were prepared in transport where  $\Delta Q/\Delta t$  is the linear rate of appearance of solute in the medium and trace amounts of <sup>14</sup>C-mannitol (DuPont Ltd, Herts, receiver solution, A is the cross-sectional area of the polycarbo-UK) added to each solution (to give a final activity of 0.5  $\mu$ Ci/ mL). At the start of each transport experiment, half of the of solute in the donor compartment at  $t = 0$ . The  $P_{app}$  values

purified on discontinuous Percoll<sup>®</sup> gradients. The purified cells replaced with an equal volume of peptide-containing medium, fluorescence assays) and 50  $\mu$ L samples (for liquid scintillation counting) were taken from the receptor compartment and sides of the monolayer was changed on alternate days. tion by HPLC, to check for peptide breakdown products. Trans-Transport experiments were carried out using a modified epithelial resistance across the monolayers was monitored from

mM NaH<sub>2</sub>PO<sub>4</sub>.2H<sub>2</sub>O, 5.55 mM *D*-glucose, 0.075 M bovine of the layers was routinely checked by monitoring their transepi-

 $(P_{app}, \text{ in cm.s}^{-1})$  were calculated as:

$$
P_{app} = (\Delta Q/\Delta t)/A \cdot c(0)
$$

nate filter (i.e., 1.13 cm<sup>2</sup>) and  $c(0)$  is the initial concentration medium from the donor compartment was removed and were not corrected for the  $P_{app}$  of the matrix-free filters (because

filter), nor were they corrected for aqueous boundary layer effects (since these are only significant for small non-electrolytes) (9). **RESULTS**

The fluxes of the peptides across the alveolar cell mono-<br>layers were routinely determined in the apical to basolateral<br>(AB) direction and, in most cases, also in the basolateral to<br>monolayers with *TER* in the range 1050– (AD) direction and, in most cases, also in the basolateral to<br>apical (BA) direction, using an initial concentration of peptide<br>in the donor chamber of 0.2 mM or 1 mM. Quantitation of the sensitive indicator of their tight

Liquid scintillation counting was used to determine  $^{14}$ C different peptides tested.<br>
Material and the issue of the mechanism of alveolar mannitol levels in samples 5 mL of Ready-Protein® scintillant and addressing firs

## **Modelling of the Peptide Permeability Data**

The potential numbers of hydrogen bonds formed between the peptides and water  $(N_h)$  were calculated by summing the appropriate group contributions, taken as:  $-OH$ , 2;  $-NH_2$ , 2;  $-C = 0$ , 2;  $-NH_3^+$ , 3;  $-COO^-$ , 4. Stokes radii of the peptides were calculated from their molecular volumes, obtained by summation of group volume increments taken from Chothia (10): Ala, 92  $\tilde{A}^3$ ; Asp, 125  $\tilde{A}^3$ ; Gly, 66  $\tilde{A}^3$ ; Ser, 99  $\tilde{A}^3$ ; Val, 142  $\AA^3$ ; Lys, 171  $\AA^3$ . The molecular volume for DCA (390  $\AA^3$ ) was calculated from density and MW data taken from the literature (11): coumarin, 260  $\AA^3$ ; methoxy substituent on aromatic ring, 32  $\AA$ <sup>3</sup>; peptide main chain (Gly), 66  $\AA$ <sup>3</sup>.

Modelling of the alveolar cell monolayer *Papp* data was **Example in the Transeptible late resistance / ohm.cm<sup>2</sup><br>performed using eqs. (1) and (2), with the peptide molecular<br>radii and calculated diffusion coefficients provided as input,<br>and the sum of the squared differences b** routine, adapted from Bunday (12). For the series III peptides,  $\phi_{\text{min}}$  ohm.cm<sup>2</sup> to 2650 ohm.cm<sup>2</sup>.

emission wavelengths of 345 nm and 445 nm, respectively (7). consideration of such a restricted data set served to minimize<br>the likelihood of significant inter-experiment variations in the **Cuantitation of Permeants** ingful comparisons could be made between the *P<sub>app</sub>* for the *P<sub>app</sub>* for the

was added to each 50  $\mu$ L sample and radioactivity measured peptide transport, experiments were conducted to see how the using a Rackbeta LKB 1209 Liquid scintillation counter (Rack-<br>using a Rackbeta LKB 1209 Liquid scin  $_{ac}X^*N_{2}$ , and the tripeptide,  $_{ac}X^*AS_{N}$ . We find, however, that **Peptide Partitioning Experiments** (at the concentrations tested) the AB and BA fluxes for these Log P (and  $logD_{7.4}$ ) determinations for the peptides were<br>carried out at 25°C using n-octanol/Krebs Ringer buffer (pH<br>7.4), with each phase carefully equilibrated with the opposite<br>phase before use. Peptide concentration



( $\varepsilon/\delta$ ) and *R*, using a Hooke & Jeeves direct search optimisation using monolayers with TER in intervals of 200 ohm.cm<sup>2</sup> from 650



across alveolar cell monolayers (data represent the means of 3 determinations  $\pm$  sem, error bars in some cases subsumed within symbol; legend shows initial concentrations of peptide). (b) Apical to basolateral (AB) and basolateral to apical (BA) flux of acX\*NH<sub>2</sub> across alveolar tions  $\pm$  sem, error bars in some cases subsumed within symbol; initial cell monolayers (data represent the means of 3 determina- peptide concentratio cell monolayers (data represent the means of 3 determina-

saturated (14–16) (together with the difficulties in rationalising why peptide transporters would be needed on the basolateral side of alveloar epithelia) we conclude that the mechanism of transport of the peptides is most likely to involve passive diffusion. There are no statistically significant differences in the AB and BA *Papp* values for four of the larger peptides studied (as judged by two-tailed t-tests; Fig. 2c) and so all of the molecules are assumed to cross the alveolar cell monolayers in the same manner.

From the data obtained here, it is not possible to draw any unequivocal conclusions as to whether the peptides traverse the monolayers through the aqueous pores provided by the epithelial fight junctions (following the paracellular route) or whether they cross by diffusing through the cytoplasm of the cells, partitioning into and out of the apical and basolateral cell membranes (following the transcellular route). It seems highly probable, however, that their major route of passage is *via* the paracellular route. Such a conclusion is based on the combined observations that all of the peptides studied partition in favour of an aqueous medium rather than an organic/lipophilic medium (with negative  $logP$  values, Table 1) and have  $P_{app}$  that are consistently lower than the corresponding  $P_{app}$  measured for the hydrophilic solute, mannitol, which is widely recognised as a marker for the paracellular route (17).

With regard to peptide structure-permeability relationships, we find that the principal determinant of a peptide's flux across the alveolar epithelium is its size. In part this is demonstrated by the observation that the series I peptides show a generally inverse dependence of their AB *Papp* values on molecular weight (see Table 1), ranging from 1.88 ( $\pm$ 0.07)  $\times$  $10^{-7}$  cm.s<sup>-1</sup> for acX\*NH<sub>2</sub>, down to 0.41 ( $\pm$ 0.26)  $\times$  10<sup>-7</sup> cm.s<sup>-1</sup> for acX\*AS(GAS)<sub>7</sub>NH<sub>2</sub>. Matsukawa *et al.* (18) have previously reported the same trend for a range of solutes transported across alveolar cell monolayers, and the same relationship has also been noted for transport of peptides across Caco-2 cell monolayers by Adson *et al.*(19). It must be noted here, however, that since the *logP* of the peptides in series I tend to decrease with increasing chain length, it is difficult to deconvolute completely the influences of their size and hydrophilicity. Our justification, therefore, for the conclusion that it is size, more so than hydrophilicity that governs alveolar peptide flux is provided by the observations made for series II peptides. With these three pairs of peptides a single Ser to Val substitution leads to a small *increase* in mass of 12, but causes a marked *decrease* in the peptide's hydrophilicity (that is, an increase in its *logP*; Table 1). Since the Ser to Val substitutions in all three lengths of peptides seem to result in insignificant differences in the peptide's AB *Papp*, it would thus appear that the change in size of the peptides in series I has more influence on transport rate than the change in *logP*.

If we err on the side of caution, however, and consider which physico-chemical property is most appropriate to guide in the task of peptide drug design, it is clearly best to try to predict alveolar permeability using an index that provides a **Fig. 2.** (a) Concentration dependence of the AB flux of acX\*ASNH<sub>2</sub> combined measure of peptide size *and* hydrophilicity. For such

cients (*Papp*) for AB and BA fluxes of peptides across alveolar cell monolayers (data represent the means of  $\geq 3$  determinations  $\pm$  sem).



line shown  $(-)$  is calculated over the data for peptides with MW's  $<$  1200, and the line then extrapolated beyond these data ( $\cdots$ ). and (employing the same type of calculation) obtained a mean

purposes we can use a simple count of the number of potential by Matsukawa and co-workers (18), who used alveolar cell hydrogen bonds formed between the peptide and water (the parameter  $N_h$  listed in Table 1). This index has already found utility in analyses of solute transport into animal and plant cells (20), and also in more specific analyses of peptide transport across Caco-2 cell monolayers (21). If our assumption that peptides cross the alveolar epithelium by means of passive diffusion *via* the paracellular route is correct, then the use of  $N<sub>h</sub>$  as a predictor of flux is, in any case, preferable to the use of *logP*. This is because *logP* provides a measure of the *relative* solubilities of the molecules in aqueous and organic media, which means that it is pertinent to the partitioning of drugs into and out of biological membranes, and is relevant therefore to the transcellular transport but not the paracellular transport of drugs.  $N<sub>h</sub>$ , on the other hand is related much more directly to the aqueous solubility of a molecule and so will provide a relevant measure of the ability of a peptide to traverse the water-filled pores of the epithelial tight junctions. For any small linear peptide it is also likely to be related to the molecule's hydrodynamic diameter, and this too will have a bearing on the ease of passage of a peptide through the tight junctional pores.

Figure 3 shows the combined data for the 9 peptides in series I and II, with their  $log(P_{app})$  values plotted against  $N_h$ . We see here that there is a log-linear dependence of  $P_{app}$  on  $N_h$ , for peptides with molecular weights less than  $\sim$ 1200. In the case of the one peptide studied that has a MW above this  $(acX*AS(GAS)<sub>7</sub>NH<sub>2</sub>; MW, 1998)$  there is some indication that its *Papp* is rather higher than would be predicted by this linear relation, and it may therefore be that large peptides such as this traverse the monolayer in a manner that is different from that exploited by the smaller molecules. Although it is difficult to determine precisely the nature of these transport routes, we show below that the data are best accounted for in terms of

$$
P_{\text{ann}} = (\varepsilon/\delta) \cdot F(r|R) \cdot D \tag{1}
$$

in which  $\varepsilon$  and  $\delta$  are respectively the fractional cross-sectional area and mean path length of the pores,  $D$  is the diffusion coefficient of the transported solute, and  $F(r|R)$  is the Renkin sieving function for cylindrical channels (23), obtained as:

$$
F(r|R) = (1 - r|R)^{2} [1 - 2.104(r|R) + 2.09(r|R)^{3}
$$
  
- 0.95(r|R)<sup>5</sup>] (2)

where *r* is the molecular radius of the solute.

Assuming a single population of uniformly-sized pores, and fitting to the AB  $P_{app}$  data obtained for the series I and II peptides, we obtain an effective pore radius of  $22 \text{ Å}$ , with the fractional area per unit length of these pores  $(\varepsilon/\delta)$  calculated as  $0.08 \text{ cm}^{-1}$ . This model gives a respectable fit to the  $P_{app}$ data (Fig. 4a), but is deficient in several respects. It does not **active of the findings of Adson** *et al.* (19), **Fig. 3.** log (*P<sub>app</sub>*) vs. *N<sub>h</sub>* for neutral 'model' peptides. The regression accord, for example, with the findings of Adson *et al.* (19), who used alveolar cell monolayers (of TER  $\sim$ 1200 ohm.cm<sup>2</sup>) pore radius of 5.5 Å calculated from their  $P_{app}$  data for mannitol and urea. It also does not accord with the results reported



different populations of pores in the epithelial cell layer. **Fig. 4.** (a)  $P_{app}$  vs. molecular radius for neutral 'model' peptides. The fitted curve is calculated assuming a single population of cylindrical **DISCUSSION** <sup>throws</sup> of radius of 22 Å. (b) Variation in monolayer *P<sub>app</sub>* as a function of solute radius (in Å) with the experimental data plotted as filled circles The mean effective radii of the tight junctional 'pores' (R) and the solid line showing the calculated permeabilities assuming 2 can be estimated from  $P_{app}$  data using the relation (22):<br>A, and cylindrical pores of radiu *P* isothiocyanate-labelled dextrans are taken from Matsukawa et al. (18) and their radii and diffusion coefficients calculated as: FD4, 14  $\AA$ , .s<sup>-1</sup>; FD10, 19 Å, 1.29  $\times$  10<sup>-6</sup> cm<sup>2</sup>.s<sup>-1</sup>; FD20, 24 Å,  $1.02 \times 10^{-6}$  cm<sup>2</sup>.s<sup>-1</sup>; FD40, 31 Å, 0.79  $\times 10^{-6}$  cm<sup>2</sup>

monolayers (of mean TER,  $2450$  ohm.cm<sup>2</sup>) to study the transfluorophore-labelled dextrans and (again following a similar method of calculation) obtained a mean pore radius of 56 Å.

data, considering not just a single population of pores, but rather *z* is the number of charges borne on the cation or anion, and a mixture of two types of pores having differing radii and  $\Delta\Psi$  is the electrochemical potential gradient. population sizes. The best-fitted values for the radii of such Using the five anionic, cationic and zwitterionic peptides pores were obtained as 15 Å and 22 nm, with  $(\varepsilon/\delta)$  of 0.089 in series III, we computed their  $P^*$  values (using Eqs. (1–2),  $\text{cm}^{-1}$  and 0.011  $\text{cm}^{-1}$ , respectively.

accounts for all of the available data remarkably well, with the obtained a best fit value for  $\Delta \Psi$ . The results of these various fit between the measured and calculated *Papp* values (Fig. 4b) calculations are summarised in Table 2. The calculated value much improved over that shown by Matsukawa *et al.* (18). of  $\Delta\Psi$  is 18.4 mV, which is entirely consistent with the value However, our estimated size for the smaller cylindrical pores of 17.5 mV found for Caco-2 cell monolayers (19). Again like  $(15 \text{ Å})$  is still somewhat larger than was estimated by Adson the Caco-2 cell monolayers, the alveolar cell monolayers appear *et al.* (19), but this is because these authors considered only a to be cation selective (presumably because of the negativelysingle population of pores and used only the  $P_{app}$  data for two charged glycocalyx borne on the monolayers), with  $P^+/P^*$  being small solutes, mannitol and urea.  $1.41$  for the singly charged  $X^*AS(GAS)_{2}NH_2$  and 1.94 for the

types of pores as a function of solute radius, we find that for tides,  $P^{-}/P^{*}$  is calculated as 0.69 for  $aX^{*}AS(GAS)_{2}$  and 0.45 molecules of the size of tripeptides (with *r* of  $\sim$  5 Å) 40% of for acX\*AS(GAD-D)<sub>2</sub>NH<sub>2</sub>, whist the zwitterionic peptide, the transport is through the 22 nm pores, and 60% through the  $acX*AS(GAD-DGAD-K)NH_2$ , seems to be transported as a neu-<br>15 Å pores. With hexapeptide-sized solutes (with r around 6 tral molecule with  $P^{\pm} \sim P^*$ . 15 A pores. With hexapeptide-sized solutes (with *r* around 6  $\AA$ ) the fluxes through the 15  $\AA$  and 22 nm pores are more or Finally, we may consider how these *in vitro* data correlate less equal. with our expectations for the permeabilities of peptides across

morphology of the alveolar cell monolayers, it is important to *vitro* model of peptide transport is very similar to the model note (as mentioned by Patton; ref 1) that pore size calculations developed by Berg *et al.* (27). On the basis of their work with of this sort, in truth only really provide a way of modelling the solute transport across isolated, perfused rat lungs, these authors *Papp* data *mathematically*, and the 'pores' considered may not showed that in the intact rat lung, the transport of a set of 11 have any physical counterparts. However, if we equate  $\delta$  with hydrophilic solutes could be modelled assuming a mixture of the span of the (typically, 3–5) tight junctional strands of the two types of cylindrical pores/channels, of radii 5 Å and 34 Å, zona occludens (24), with each strand being composed of pro- with the smaller pores accounting for roughly 99% of the total tein sub-units of mean diameter 8 nm (25), we obtain a value pore area. Since the solutes included in the *in vivo* experiments for  $\delta$  in the range 25–40 nm. With this estimate for  $\delta$  we can vere all smaller than 6 Å radius, one cannot attach any great then use our calculated values for *R* and  $\varepsilon/\delta$  to determine the significance to the fact that there was no evidence found for a densities of each type of pore in the monolayer. By this means population of very large pores, and if in fact we calculate the we find that there are roughly 3.2 million 15 Å pores and 2.4 mean radius of our small and large cylindrical pores, weighted thousand 22 nm pores per cm<sup>2</sup>. Now, given that the mean in accordance with their relative areas, we obtain a 'single pore' surface area of an alveolar (type II) cell is roughly 60  $\mu$ m<sup>2</sup> size estimate of 38 Å, which is close to the value of 34 Å  $(26)$ , then there will be around 1.6 million cells per cm<sup>2</sup>. The determined by Berg *et al.*  $(27)$ . number of tri-junctional complexes will thus be roughly 3.2 As further support for our *in vitro* model of alveolar epithemillion per cm<sup>2</sup> (i.e., an average of 1 junction per cell, assuming lial cell transport, we take our two-pore model of the alveolar a hexagonal array of cells). We note, therefore, that our calcu- cell monolayers and compare the calculated permeabilities lated dimensions and relative areas of the pores are consistent  $(P_{calc})$  for selected solutes with the experimentally-determined with a monolayer in which all cells are separated from their first order rate constants for th with a monolayer in which all cells are separated from their neighbours by tight junctions, with the 15 Å pores representing across isolated, perfused rat lung (28). Plotting  $log(k)$  against the tri-junctional complexes, and the 22 nm channels perhaps  $log(P_{calc})$  (Fig. 5), we find an extremely good linear relation representing sparse holes or defects in the monolayers. (Alterna-<br>(with a correlation coefficient of representing sparse holes or defects in the monolayers. (Alternatively, we might view the very large pore transport as the uptake of solutes *via* caveolae.)

$$
P^- = P^* \cdot (\kappa/e^{-\kappa} - 1) \tag{3}
$$

$$
P^+ = P^* \cdot (\kappa/1 - e^{\kappa}) \tag{4}
$$

size-restricted diffusion of the molecules (i.e., the permeability administered *via* the pulmonary route, even when they lie within

coefficient for the neutral image of the charged species) and  $\kappa$ port of a range of solutes including high molecular weight is the dimensionless electrochemical energy function given as:

$$
\kappa = e \cdot |z| \cdot \Delta \Psi / kT \tag{5}
$$

In view of these deficiencies, we elected to re-model the where *kT* is the thermal energy, *e* is the unit charge of an ion,

assuming the two pore model of the monolayer), and then using This final two-pore modelling is eminently satisfying, and Eqs. (3–5), along with the charged peptides' *Papp* values, we If we calculate the relative fluxes through the two different doubly charged  $aX^*AS(GA \cdot D-K)_2NH_2$ . For the anionic pep-

As to the significance of these findings in regard to the the alveolar epithelia *in vivo*. We note first of all that our *in*

$$
log(k) = 0.87(\pm 0.08)log(P_{app}) + 5.78(\pm 0.53)
$$
 (7)

Moving on to the series III peptides, to consider the effects Although this correlation is based on data for just seven different of charge on alveolar peptide transport, we have that (22): solutes, it is nevertheless encouraging given that the data are distributed fairly evenly along the fitted line, with the solutes considered having sizes covering one order of magnitude.

for anionic species, and: We conclude that the monolayers prepared as primary cultures of rat alveolar cells provide a reliable way of predicting the *in vivo* transport properties of solutes, and that peptides (as for cationic species, where *P*\* is the permeability coefficient for indeed any other class of compound) can be very effectively





*Note:* r, molecular radius; z, molecular charge; D, diffusion coefficient (calculated using the Stokes-Einstein equation); P<sub>app</sub>, measured mean apparent permeability coefficient for transport of the peptide across rat alveolar cell monolayers;  $P^{\pm}$ , calculated permeability of the charged peptide; P\*, calculated permeability for the neutral image of the peptide

the macromolecular size range. On the basis of the work GlaxoWellcome), Ciba Geigy (now Novartis), ICI Pharmaceutireported here we consider that all peptides of more than three cals (now Zeneca) and SmithKline Beecham. residues will cross the alveolar epithelium *via* the paracellular route; those with six residues or fewer passing through the tri-<br>junctional complexes; and those that are larger mainly or wholly<br>**REFERENCES** exploiting gaps/defects in the epithelium, or alternatively trav-<br>ersing the monolayer in caveolae. Given the general desire<br>to develop peptide drugs involving six or fewer residues, the<br> $\frac{2}{D}$  A Wall Pulmonary absorpt to develop peptide drugs involving six or fewer residues, the 2. D. A. Wall. Pulmonary absorption of peptides and proteins. *Drug* pulmonary route of administration would thus seem perfectly *Delivery* 2:1–20 (1995). pulmonary route of administration would thus seem perfectly *Delivery* 2:1–20 (1995).<br>exploitable Such molecules will be able to gain access to the 3. L. Wang, D. Toledo-Velasquez, D. Schwegler-Berry, J. K. H. Ma, exploitable. Such molecules will be able to gain access to the service inculation through both the 15  $\AA$  and 22 nm pores.<br>From the results reported here it would also seem that subtle and Y. Rojanasakul. Transport and h From the results reported here it would also seem that subtle<br>changes in the hydrophobicity of peptides would probably not 4. K. Morimoto, H. Yamahara, V. H. L. Lee, and K.-J. Kim. Dipeptide changes in the hydrophobicity of peptides would probably not have too pronounced an affect on their alveolar permeability, transport across alveolar epithelial cell monolayers. *Pharm. Res.*<br>**10**:1668–1674 (1993). and that if the molecules carried a net positive charge this would<br>certainly enhance their permeability-with peptides incorporat-<br>ing imidazole groups (with  $pK_a$ 's close to body  $pH$ ) perhaps<br>able to exploit both transce able to exploit both transcellular as well as paracellular routes 6. F. M. Bennett, D. J. Barlow, A. N. O. Dodoo, R. C. Hider, A. B.<br>
Lansley, M. J. Lawrence, C. Marriott, and S. S. Bansal. L-(6,7-

with industrial contributions provided by Wellcome (now



**Fig. 5.** Correlation between the calculated alveolar cell monolayer by injection  $\text{Res}(P_1)$ , for a cleated achitector  $(P_2)$ , and their corresponding.  $\text{res}(P_1)$  $log(P_{amp})$  for selected solutes ( $P_{calc}$ , cm.s<sup>-1</sup>) and their corresponding  $\log(k)$  where *k* is the first order rate constant (in units of h<sup>-1</sup>) for their 15. M. Boll, M. Herget, M. Wagener, W. M. Weber, D. Markovich,  $\log(k)$  where *k* is the first order rate constant (in units of h<sup>-1</sup>) for thei  $\frac{1}{2}$  absorption across isolated, perfused rat lungs. The data on k are taken<br>from Schanker and Hemberger (28): urea (radius 2.7 Å), 8.85 h<sup>-1</sup>;<br>from Schanker and Hemberger (28): urea (radius 2.7 Å), 8.85 h<sup>-1</sup>;<br>proto from Schanker and Hemberger (28): urea (radius 2.7 Å), 8.85 h<sup>-1</sup>;<br>erythritol (radius 3.2 Å), 1.12 h<sup>-1</sup>mannitol (radius 3.9 Å), 0.69 h<sup>-1</sup>;<br>**93**:284–289 (1996). sucrose (radius 5.6 Å), 0.50 h<sup>-1</sup>; vitamin B<sub>12</sub> (radius 7.0 Å), 0.22 h<sup>-1</sup> inulin (radius 12.7 Å), 0.19 h<sup>-1</sup>; FD20 (radius 24.0 Å), 0.06 h<sup>-1</sup>. M. F. Romero, S. K. Singh, W. F. Boron, and M.A. Hediger,

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